

# Light and oxygen in lakes

Limnology

Lecture 7

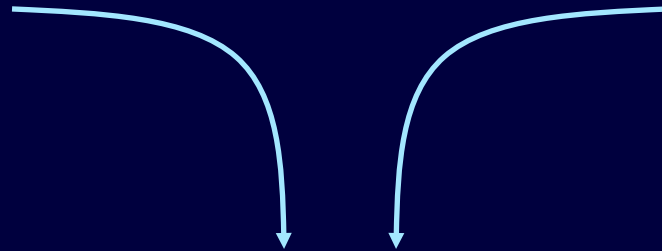
# Outline

- Light
- Diffusion
- Oxygen gradients

# Vertical Gradients in Lakes

Properties of  
Water

Properties of Lake  
Basins

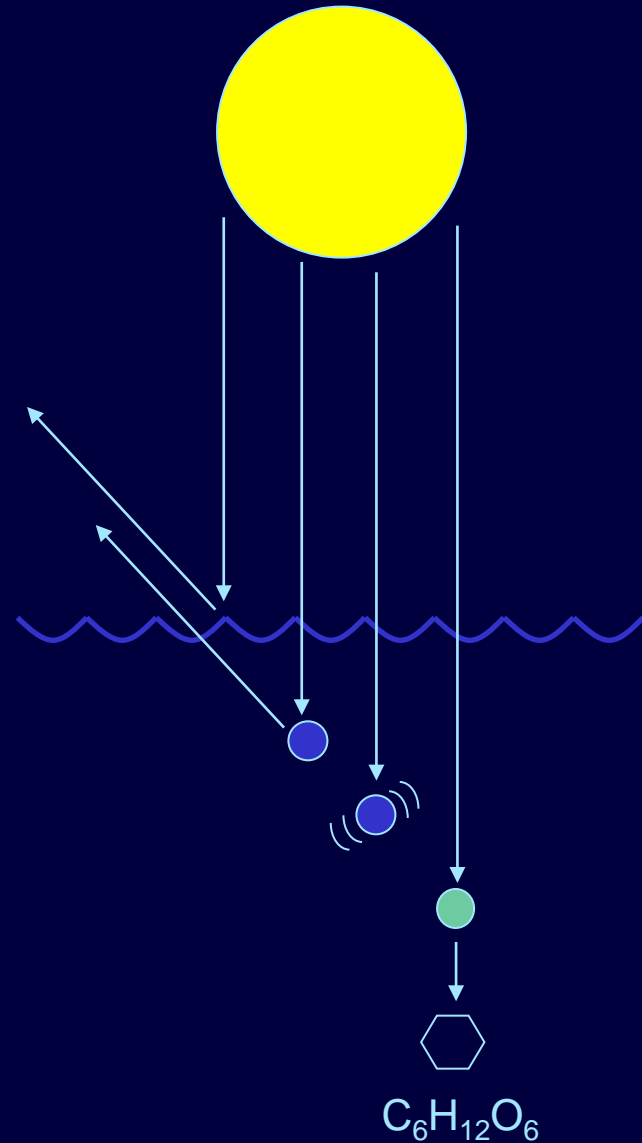


Vertical  
Gradients

- Light
- Heat
- Dissolved gases
- Dissolved nutrients

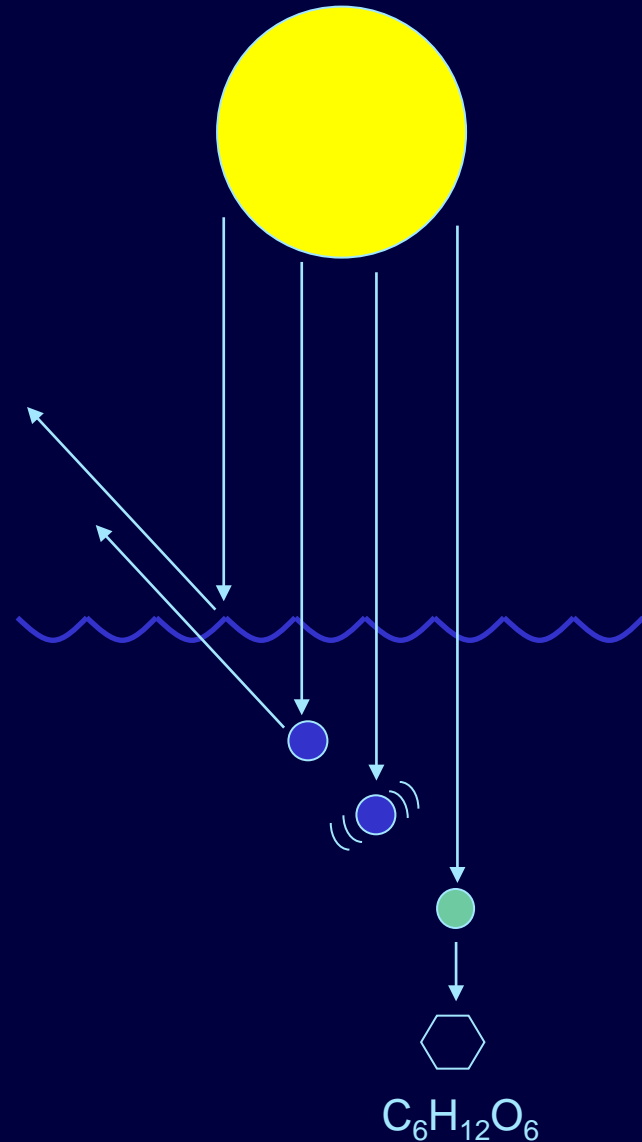
# Light and Water

- Reflection
- Absorption

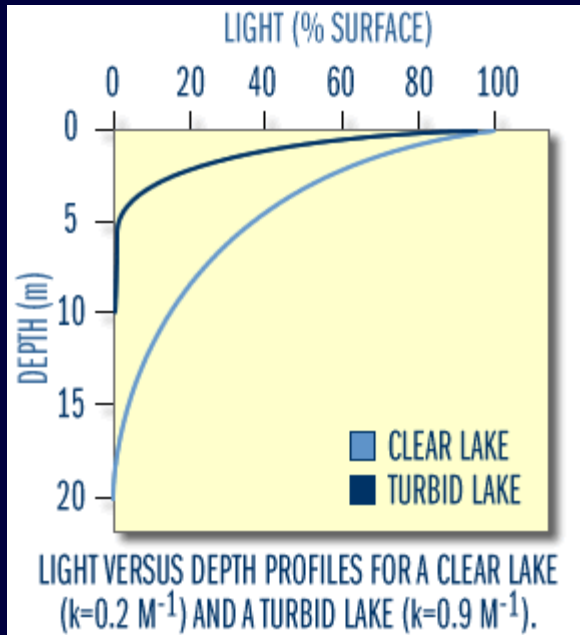


# Light and Water

- Reflection
  - Surface
  - Scatter
- Absorption
  - Heat
  - Potential Energy
    - Stored in chemical bonds via Photosynthesis



# Light and Water



Attenuation equation

$$I_z = I_0 e^{-kz}$$

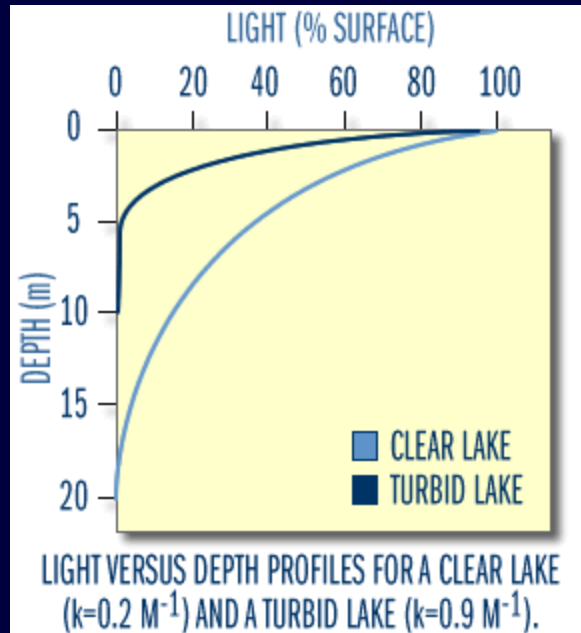
where e = natural logarithm

k = attenuation coefficient

characteristic for each water body and each wavelength

$$K = K_{\text{water}} + K_{\text{dissolved organics}} + K_{\text{particulates}}$$

# Light Gradients in Water

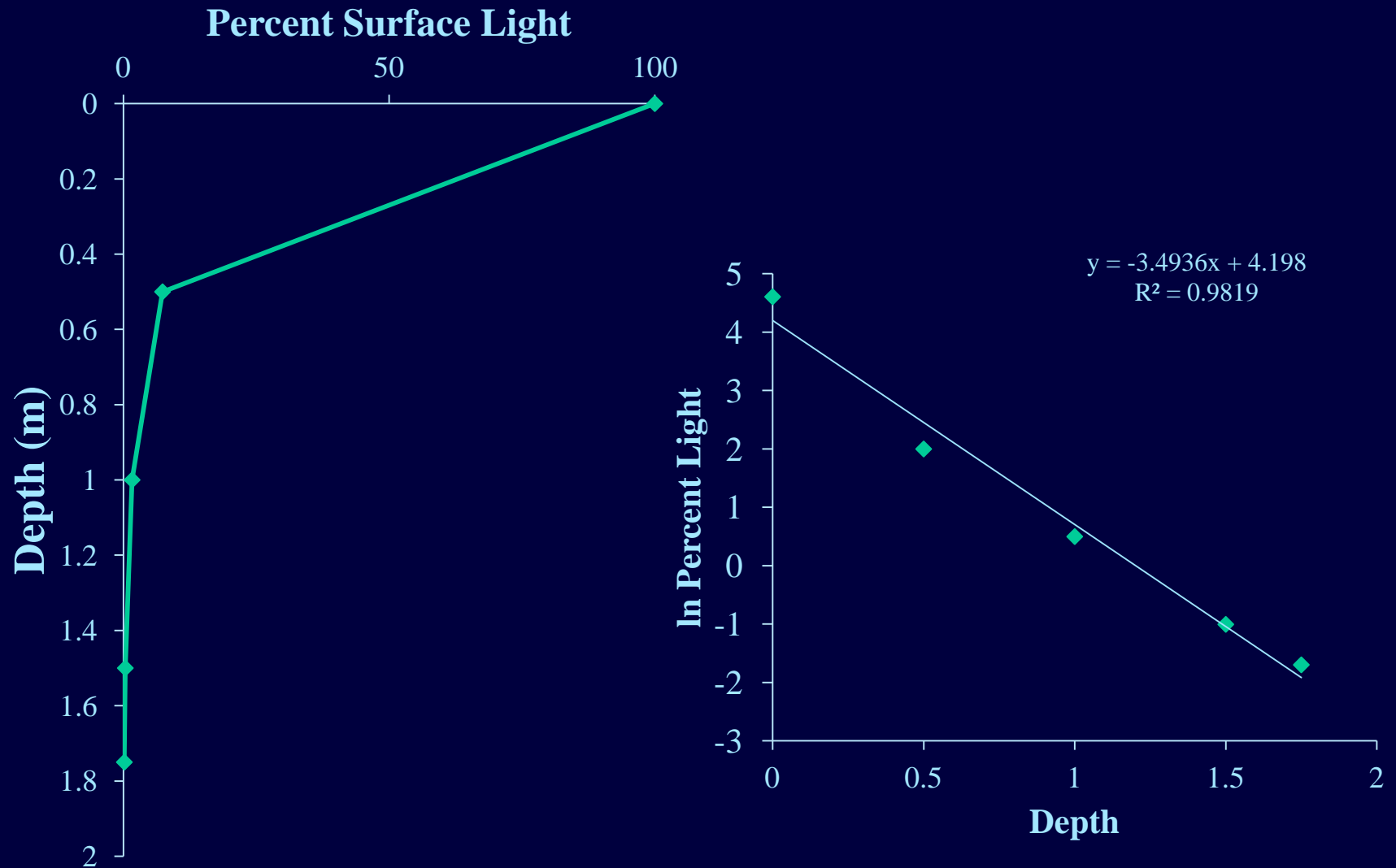


$K_{\text{dissolved organics}}$  – humic acids absorb short wavelengths (blue, UV)

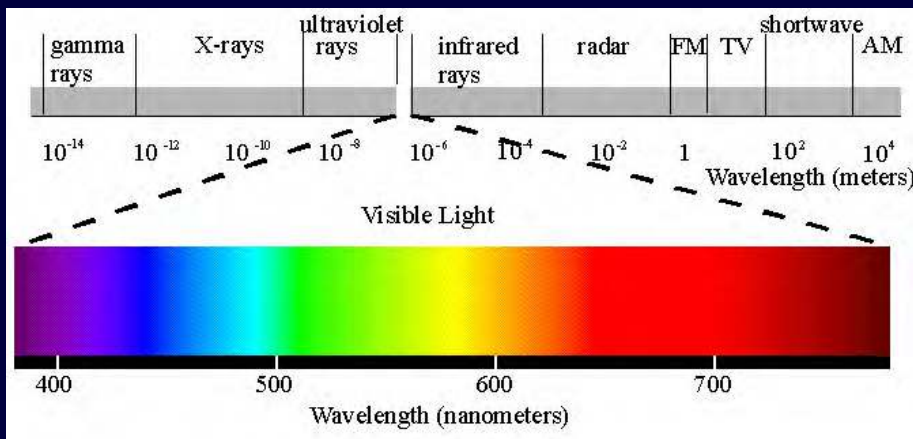
# Dunham Pond



# Dunham Pond



# Why is deep water blue?



$K_{\text{water}}$  – pure water absorbs long wavelengths

Blue absorbed the least, scattered the most

# Water decreases light and changes colors

$K_{\text{water}}$  – pure water absorbs long low energy wavelengths

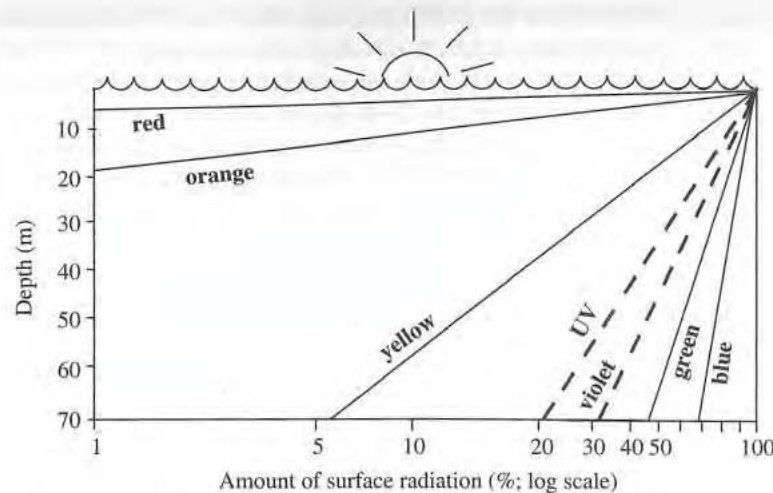
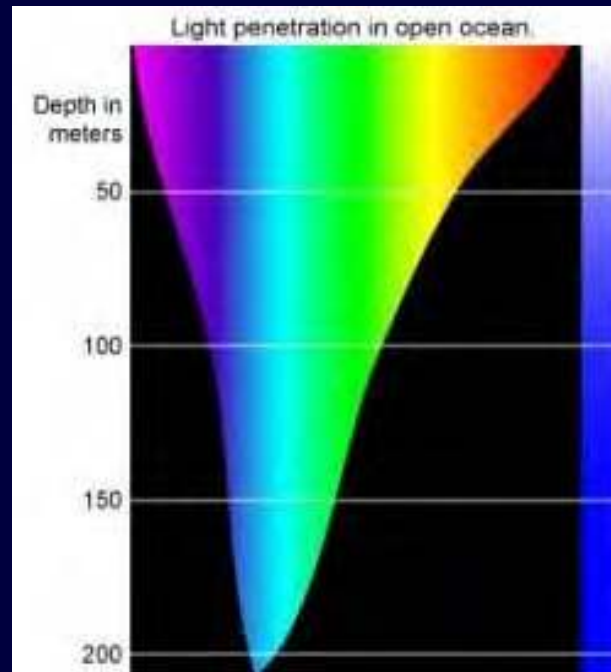
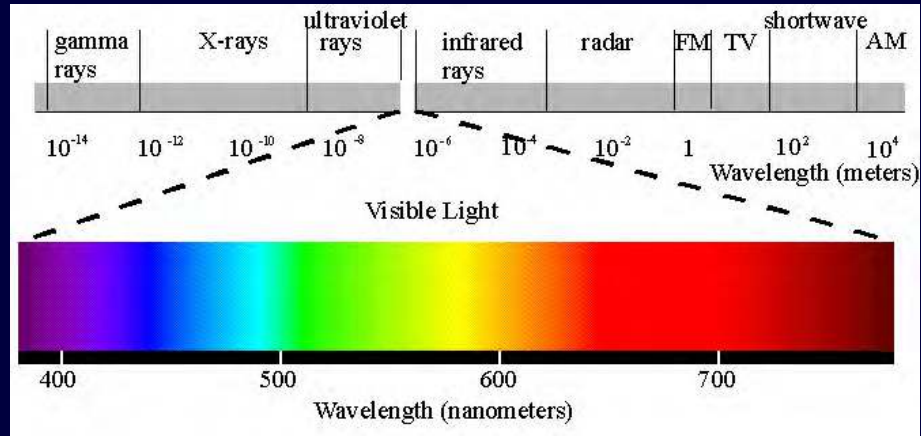


Fig. 2.9

Light transmission in distilled water showing that red light (680 nm) is already attenuated at a few metres depth, followed by orange (620 nm) reaching less than 20 m depth. About 5% of the yellow (580 nm), 46% of the green (520 nm), and almost 70% of the blue (460 nm) light still remains at 70 m depth. Note that violet and UV radiation, which have the shortest wavelengths (400 nm and <350 nm, respectively) do not reach as deep as green and blue radiation.

# Solar Radiation as a Spectrum

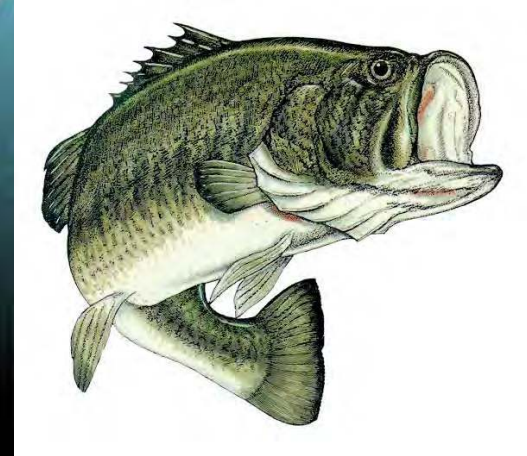
## Electromagnetic Spectrum



# Light Gradients and aquatic coloration

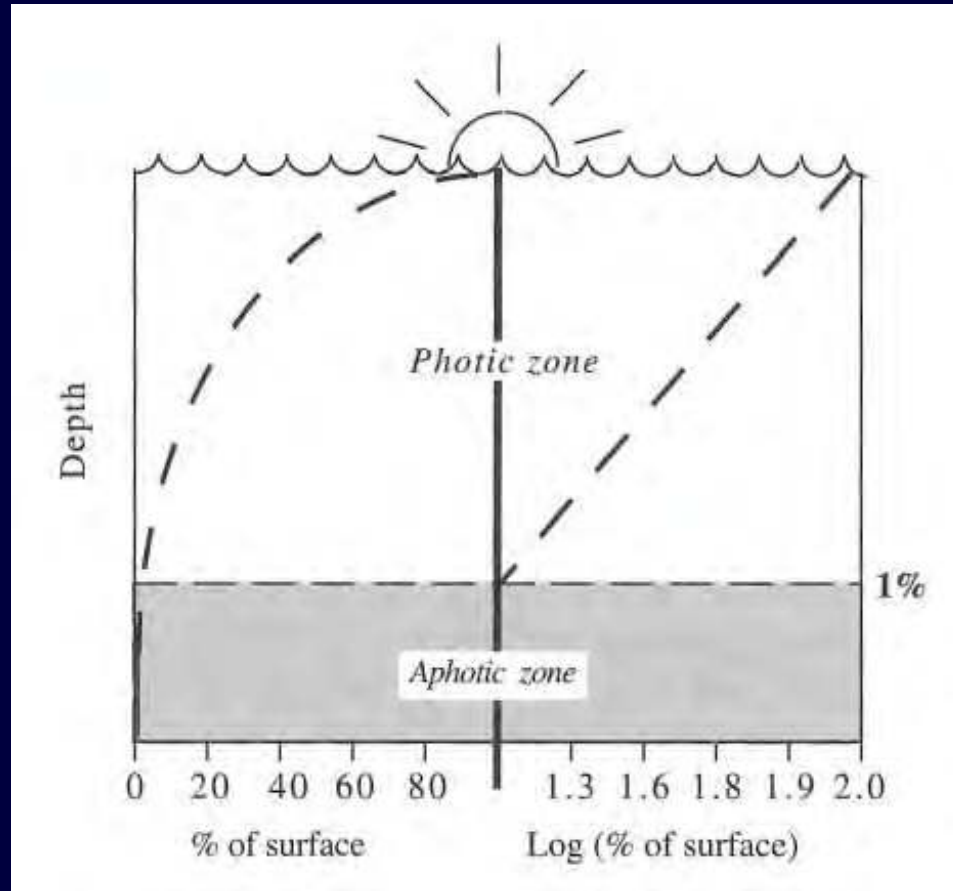
photic

twilight





# Light and habitat



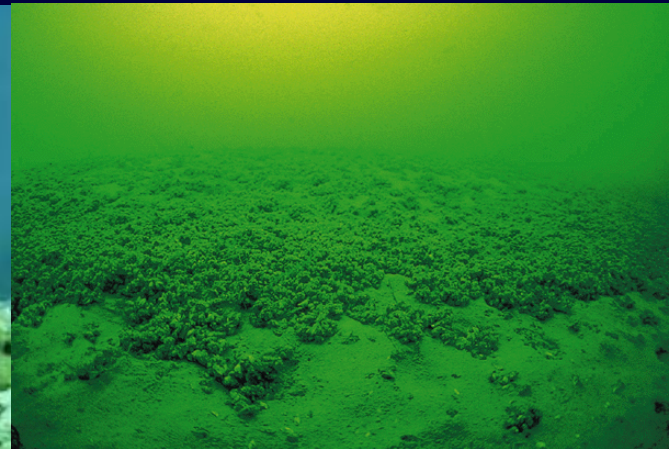
Secchi disk ~ 10% of surface light

Photic zone – light is  $> 1\%$  of surface value



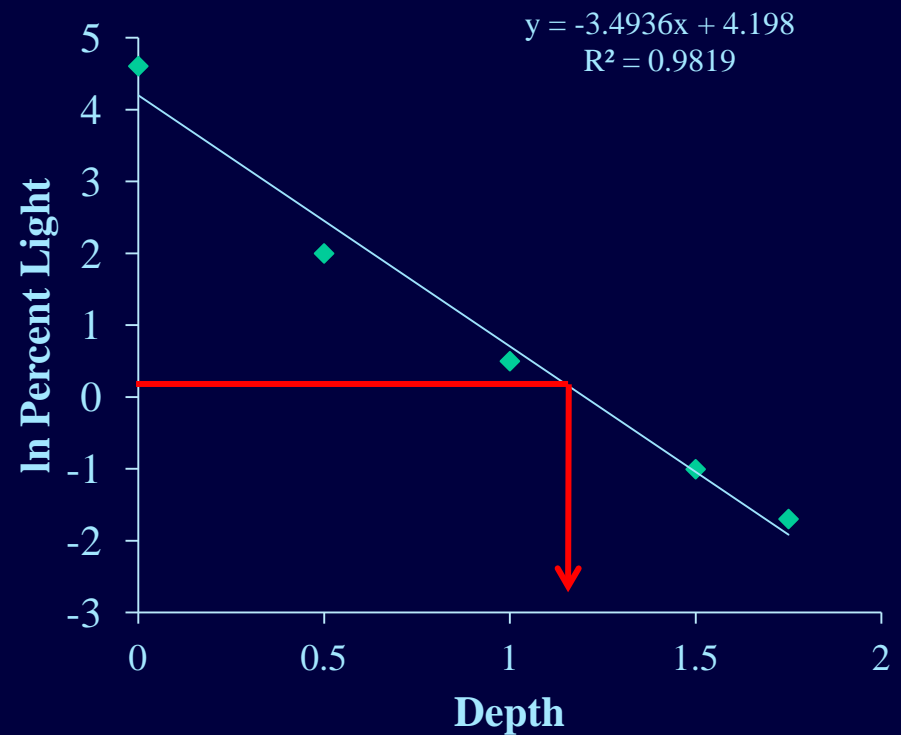
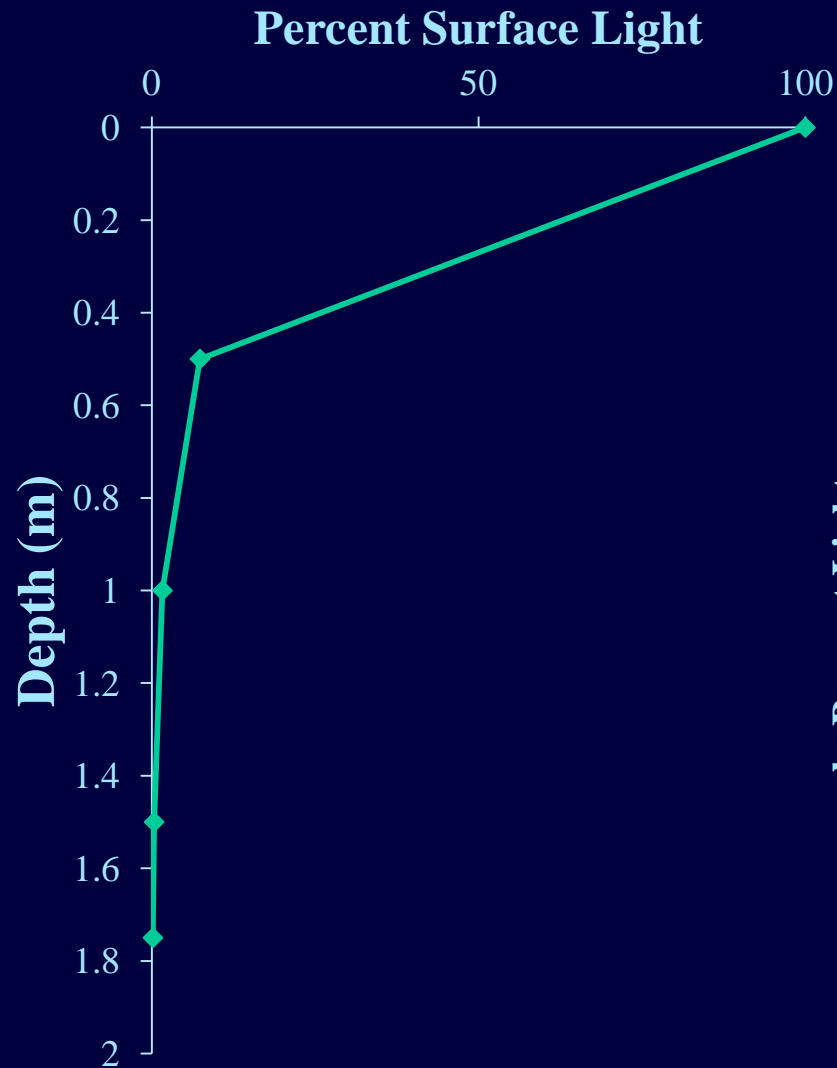
# Light Gradients in Water

Lake	k	Secchi Depth	Euphotic
Crater Lake (OR)	0.06—0.12	25— 45	>120
Lake Baikal	0.2	5— 40	15—75
Lake Erie	0.2 — 1.2	2—10	12—26
Dunham Pond	3.5	?	?





# Dunham Pond



# Light and habitat

# Light and habitat

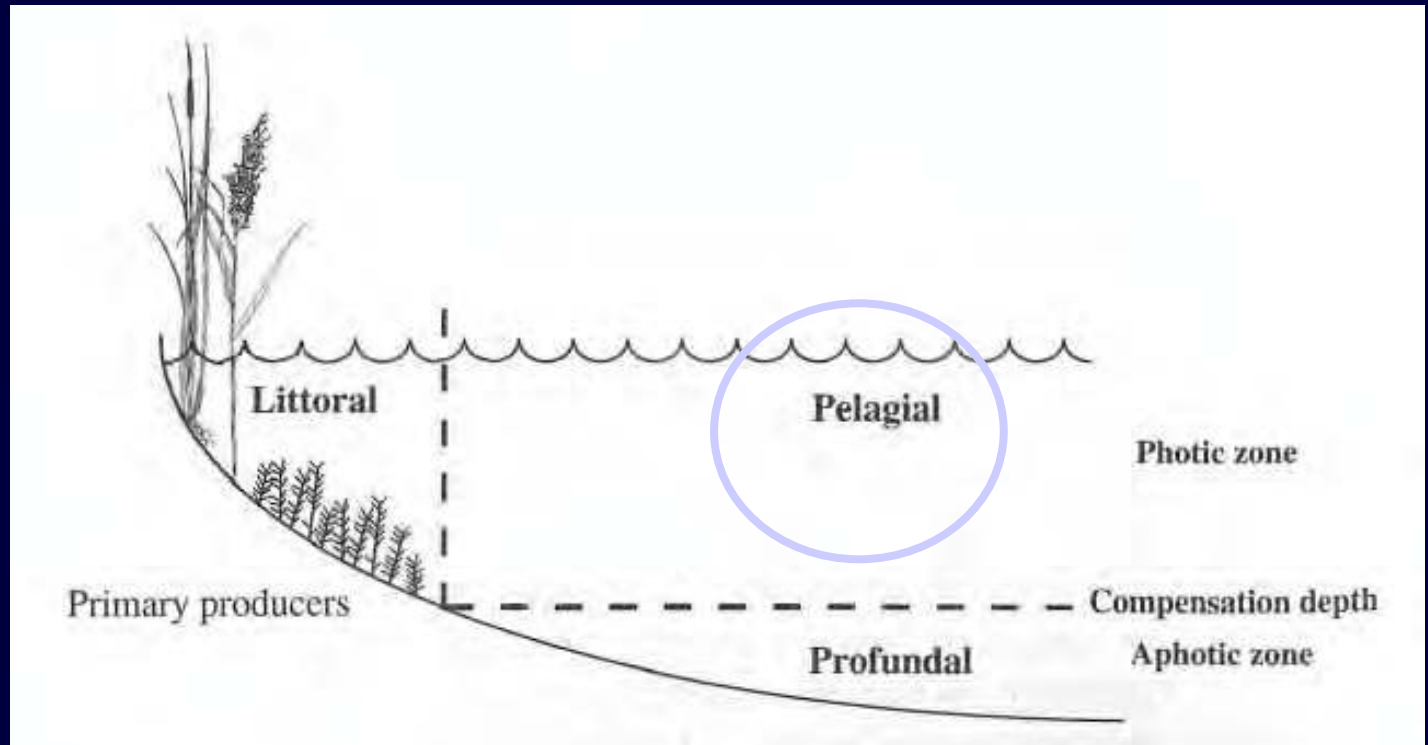


Compensation depth – where photosynthesis = respiration in plants

Littoral – From shore to aphotic zone  
- emergent and benthic plants

Limnetic – aphotic zone on benthos

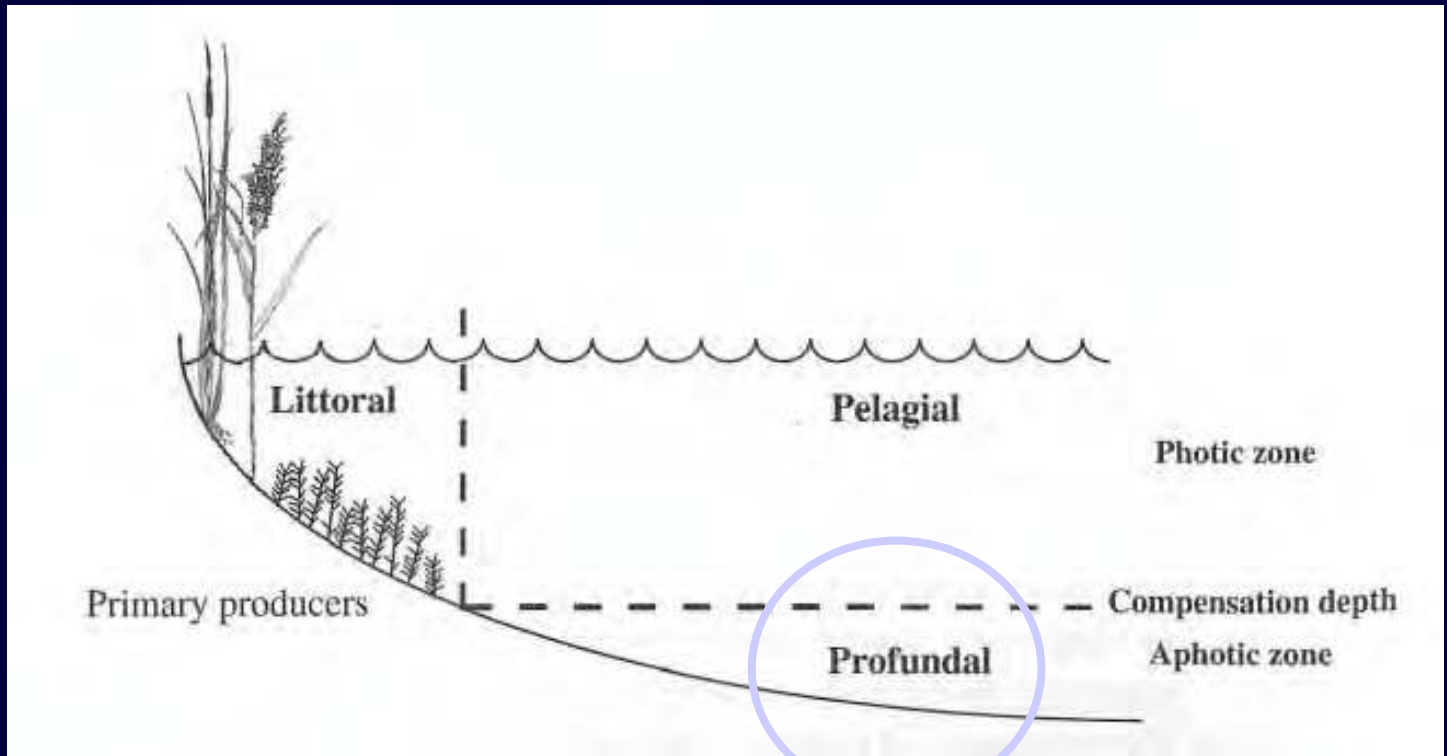
# Light Gradients in Water



Pelagic – open water in limnetic zone

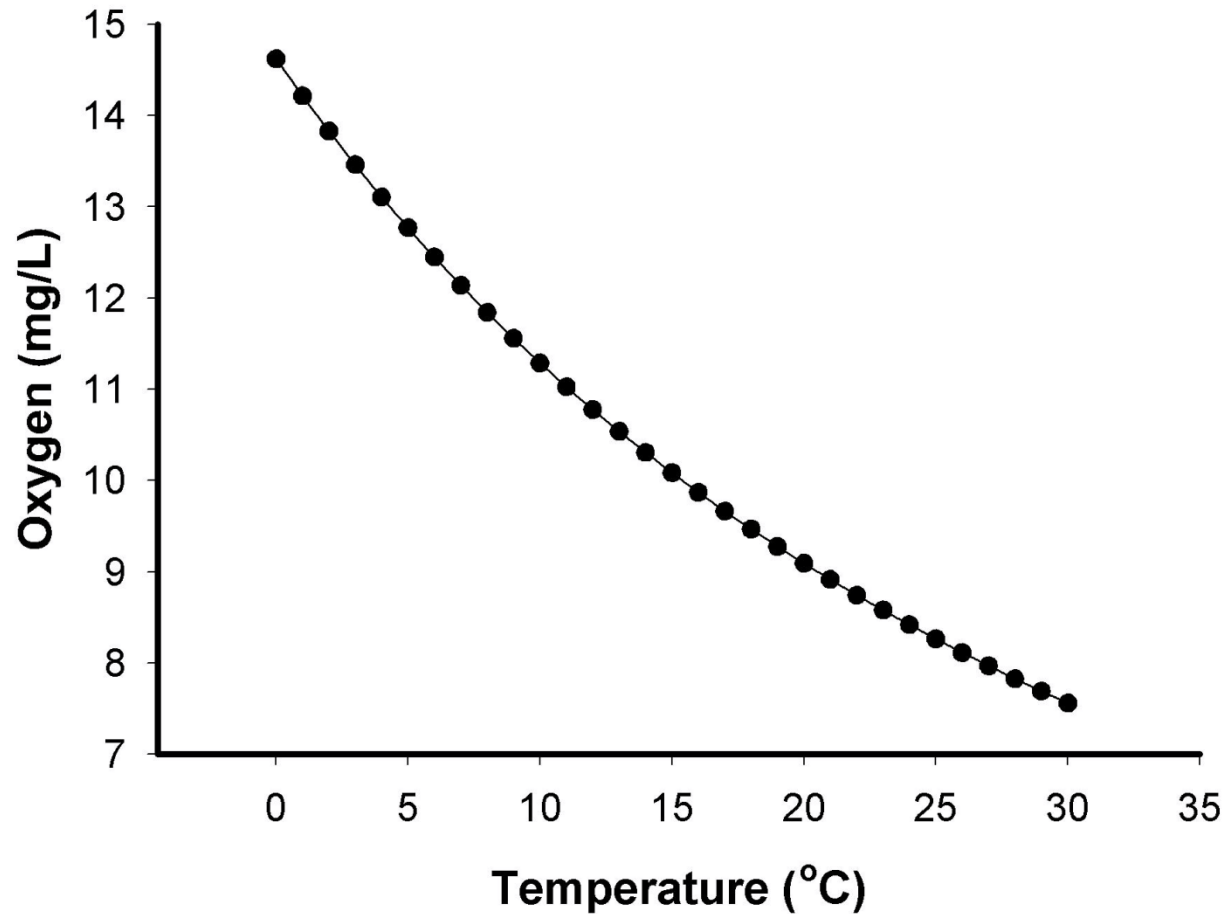
Compensation depth – where net photosynthesis = 0  
photosynthetic production = respiration  
photic or euphotic/aphotic

# Light Gradients in Water



# Oxygen and temperature

Solubility of oxygen with temperature



# Diffusion Equilibrium

Henry's Law:

$$C_s = K_H P_t$$

$C_s$  = amount of gas dissolved

$K_H$  = solubility coefficient for a given  
temperature

$P_t$  = partial pressure of gas in atmosphere

# Oxygen in water

Diffusion from atmosphere

O<sub>2</sub> partial pressure = 0.203 atm

K<sub>H</sub> (20C) = 1.39 mmol O<sub>2</sub>/kg H<sub>2</sub>O x atm

$$C_s = K_H P_t$$

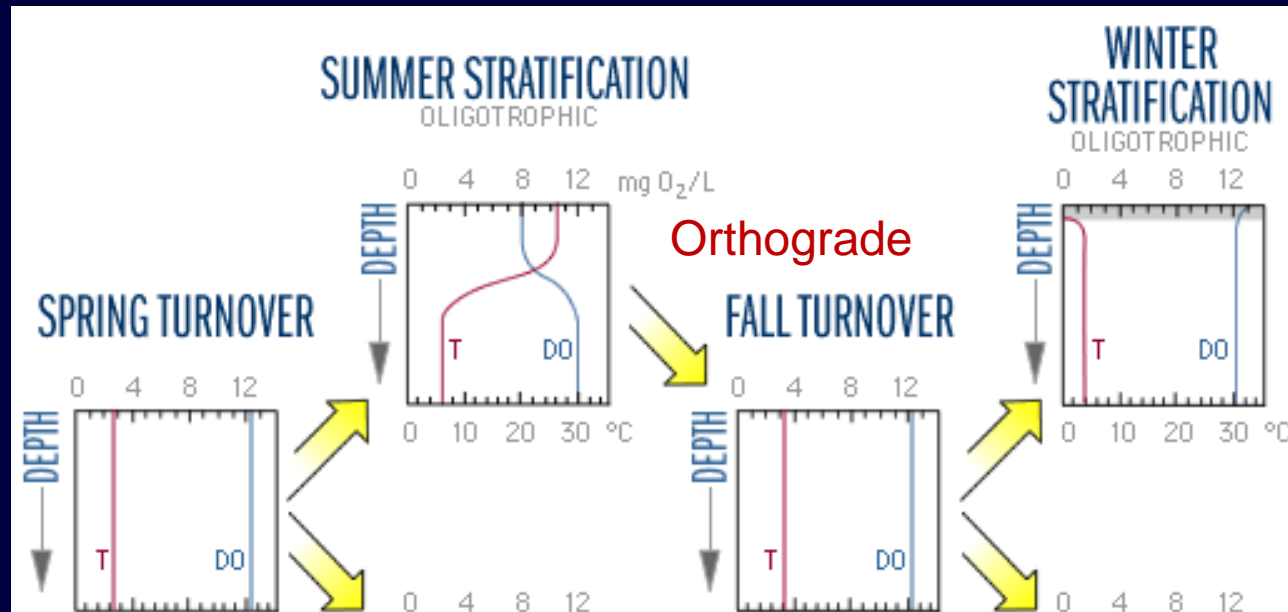
$$\begin{aligned} \text{O}_2 (20\text{C}) &= 1.39 * 0.20 \\ &= 0.28 \text{ mmol/kg} \\ &= 9.03 \text{ mg/L} \end{aligned}$$



# Oxygen and species survival

Species	DO limit (mg/L)
Trout	7-8
Bass	5
Sunfish	4.7
Carp	4
Amphibian larvae	1-2
Amphipods	2
Chironomids	1
Worms	0.7

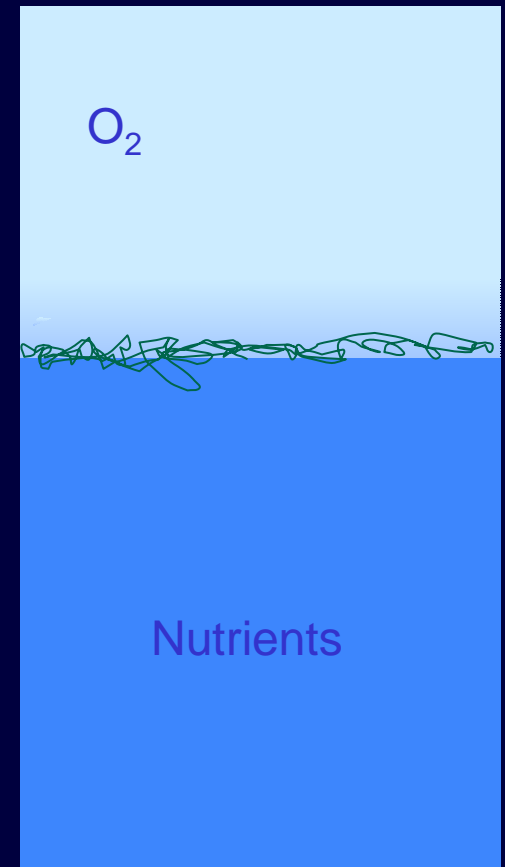
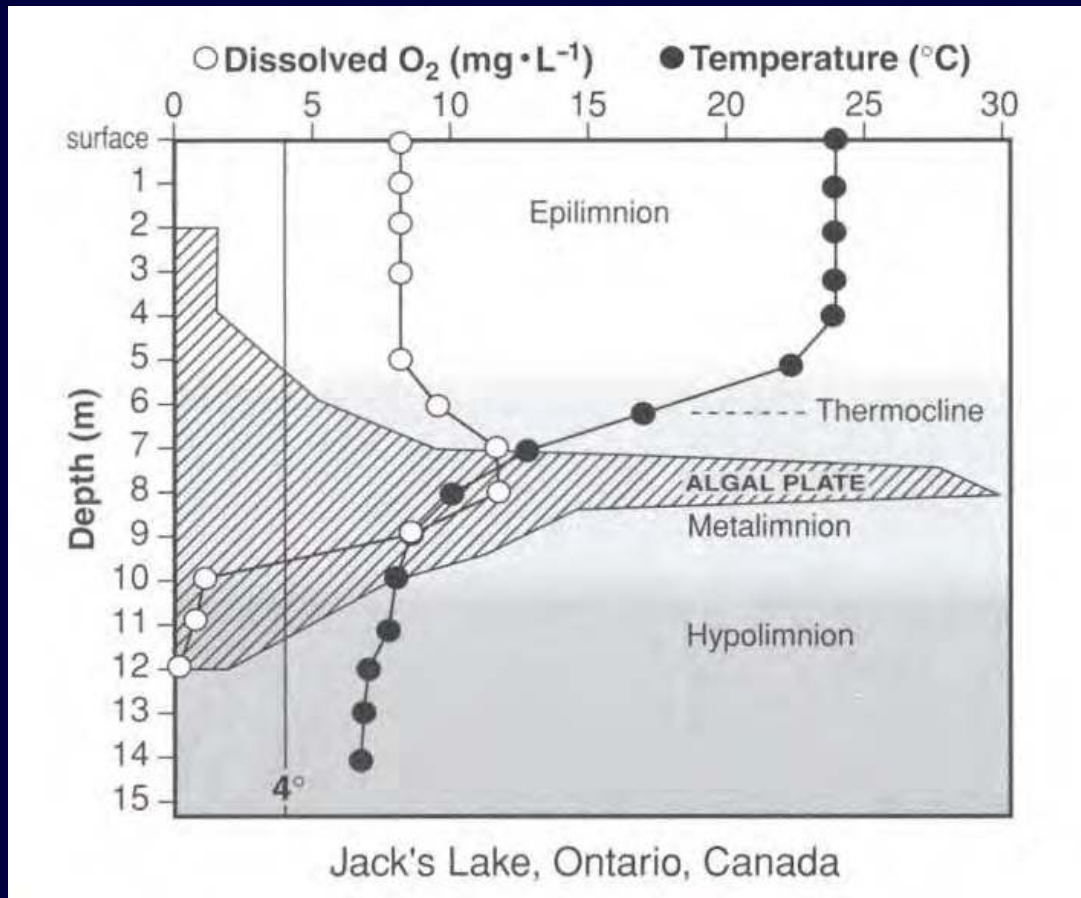
# Oxygen Gradients in Dimictic Lake



Biological oxygen demand

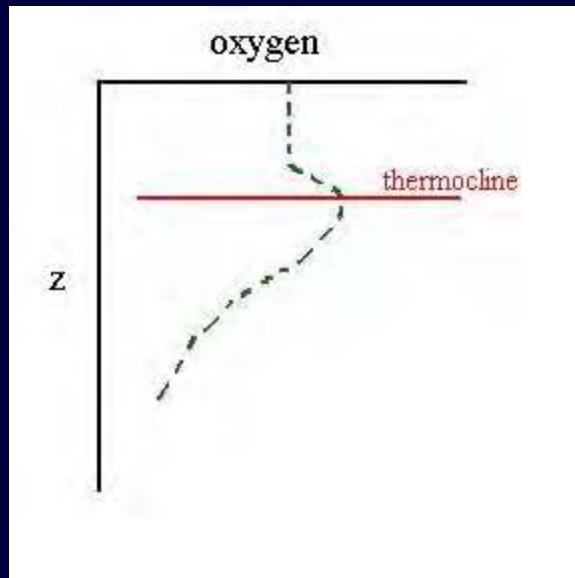
-Rate of oxygen uptake  
by aquatic organisms

# Thermocline effects



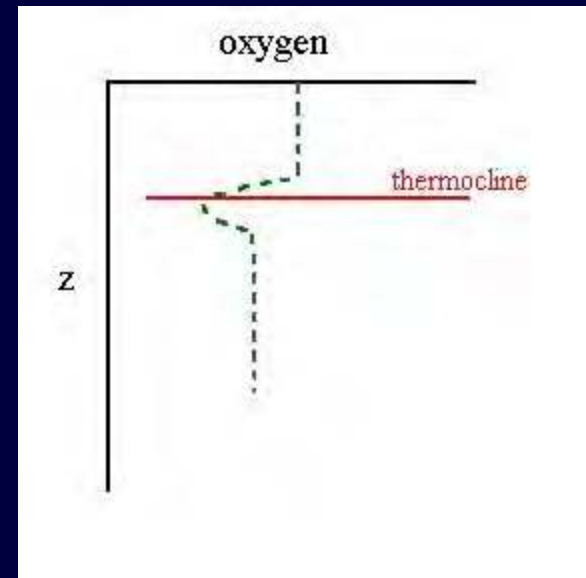
# Dunham oxygen

# Oxygen Gradients in Dimictic Lake



Positive heterograde

Algae sit on density “shelf”  
More nutrients



Negative heterograde

Respiration of algae, zooplankton,  
decomposition of detritus rain on  
“shelf”

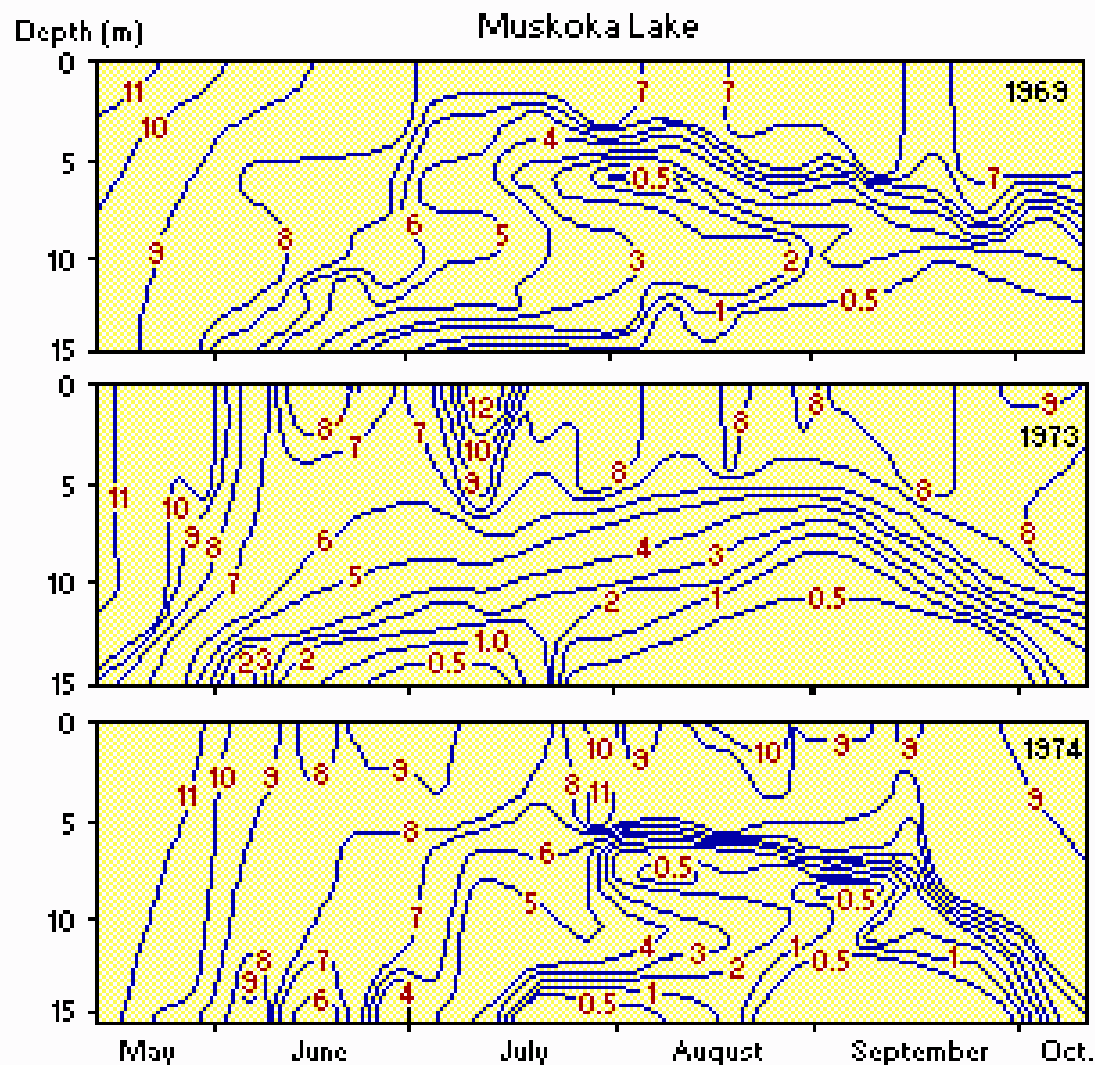


Fig. NAM-24-5 Seasonal isopleths of dissolved oxygen [ $\text{mg l}^{-1}$ ] in Gravenhurst Bay based on weekly measurements, May-October 1969, 1973 and 1974 (10).

# Fish Kills (low O<sub>2</sub>)



Occur often in small ponds in winter  
under ice cover – why?

## The winterkill process.

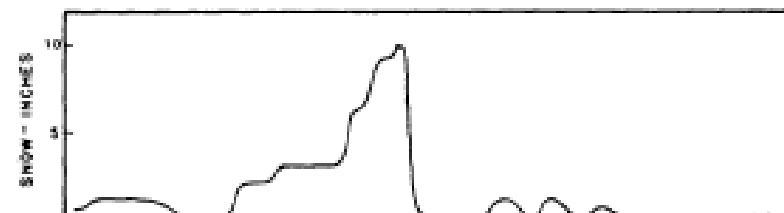
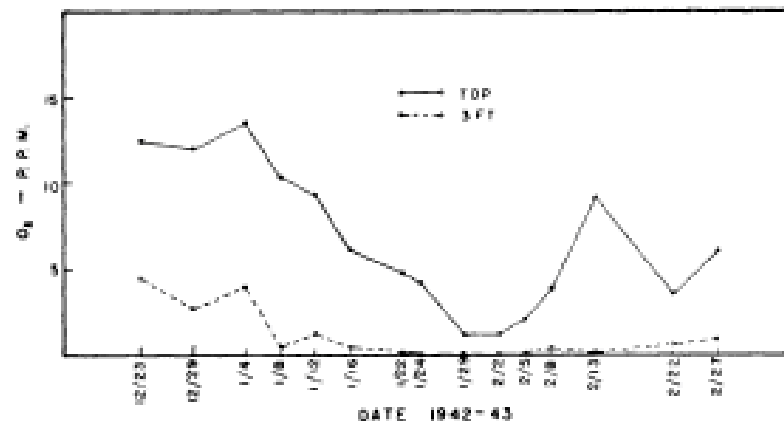
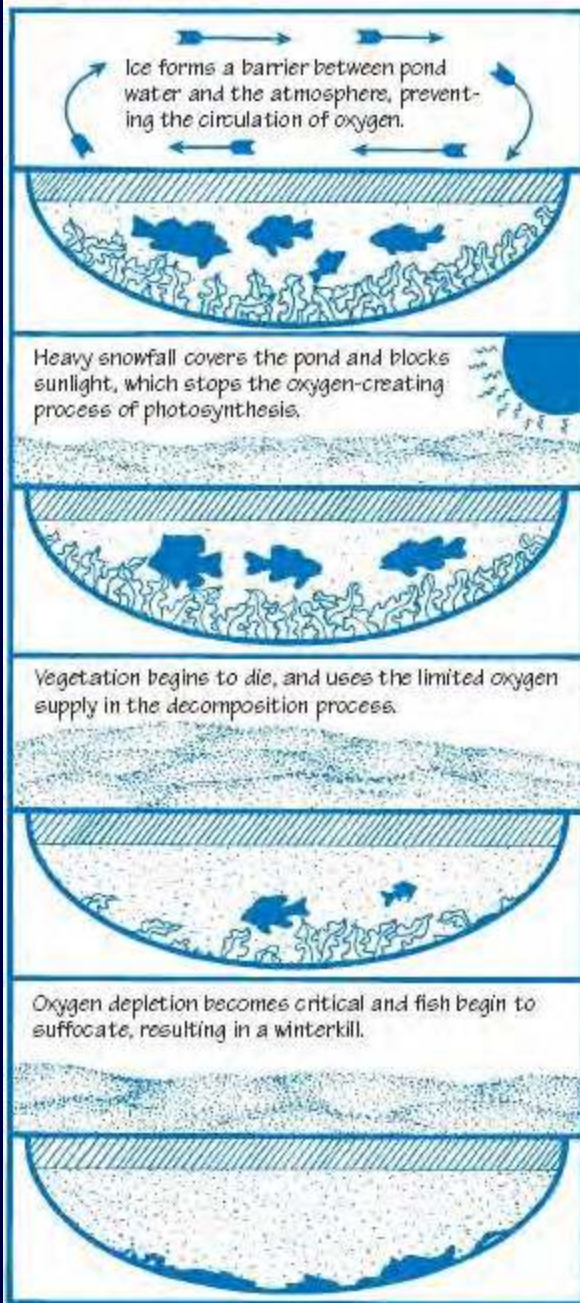


FIG. 15. Dissolved oxygen and snow cover, Green Lake, Station 1, 1940-41 and 1942-43.

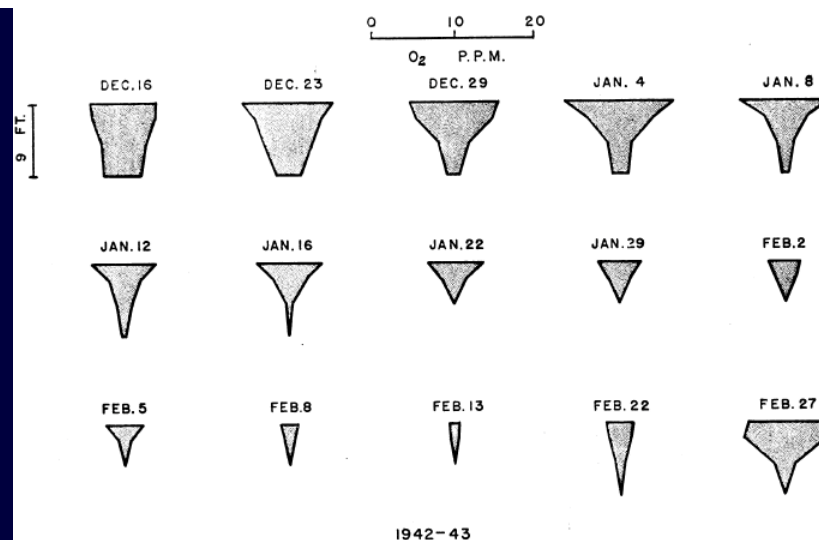


FIG. 21. Dissolved oxygen, vertical distribution, Green Lake, selected dates, 1942-43. The width of the block diagram at any given depth indicates the amount of oxygen present at that depth.



# Fish Kills (low O<sub>2</sub>)



Also occur often in large lakes in late summer  
– why?

# Late summer

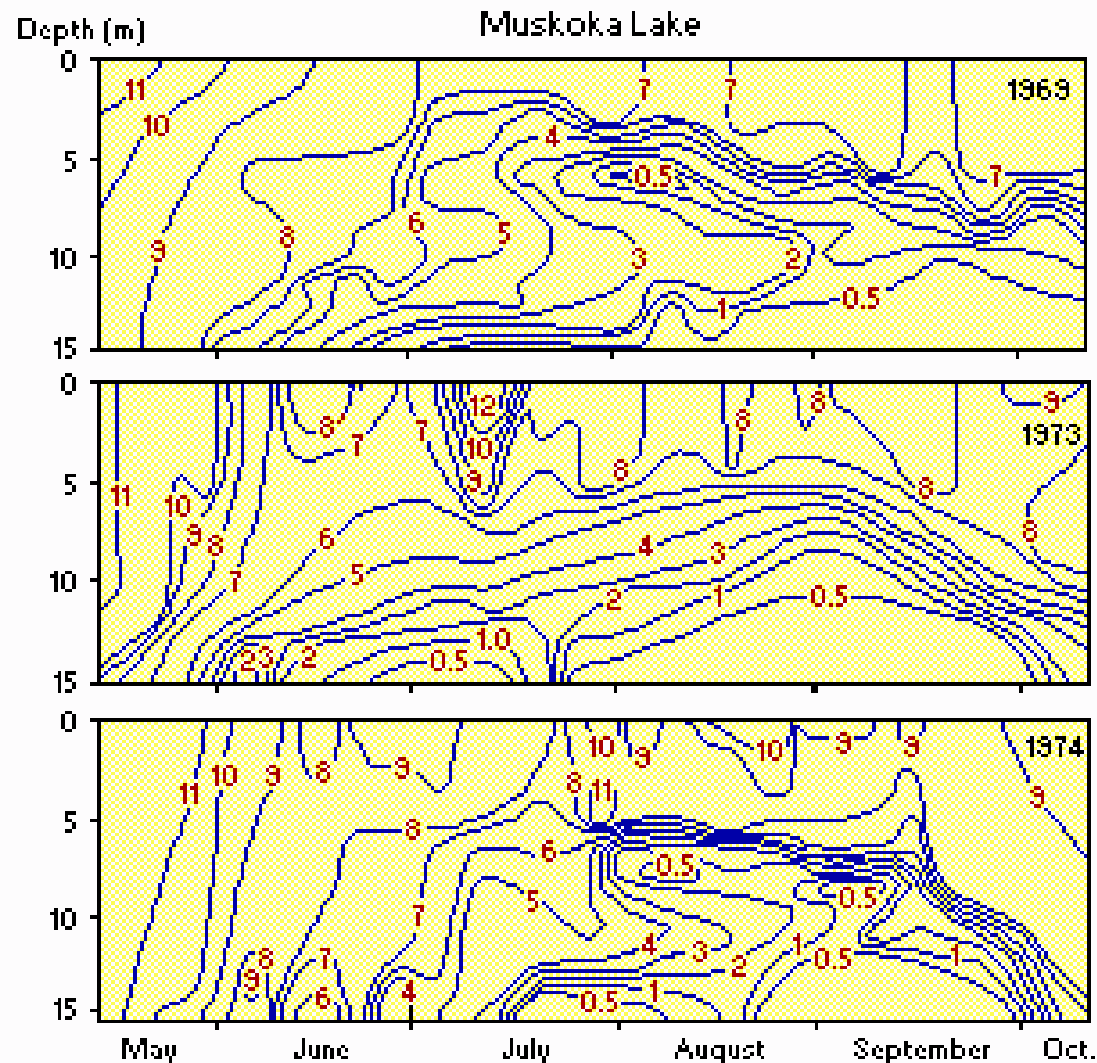


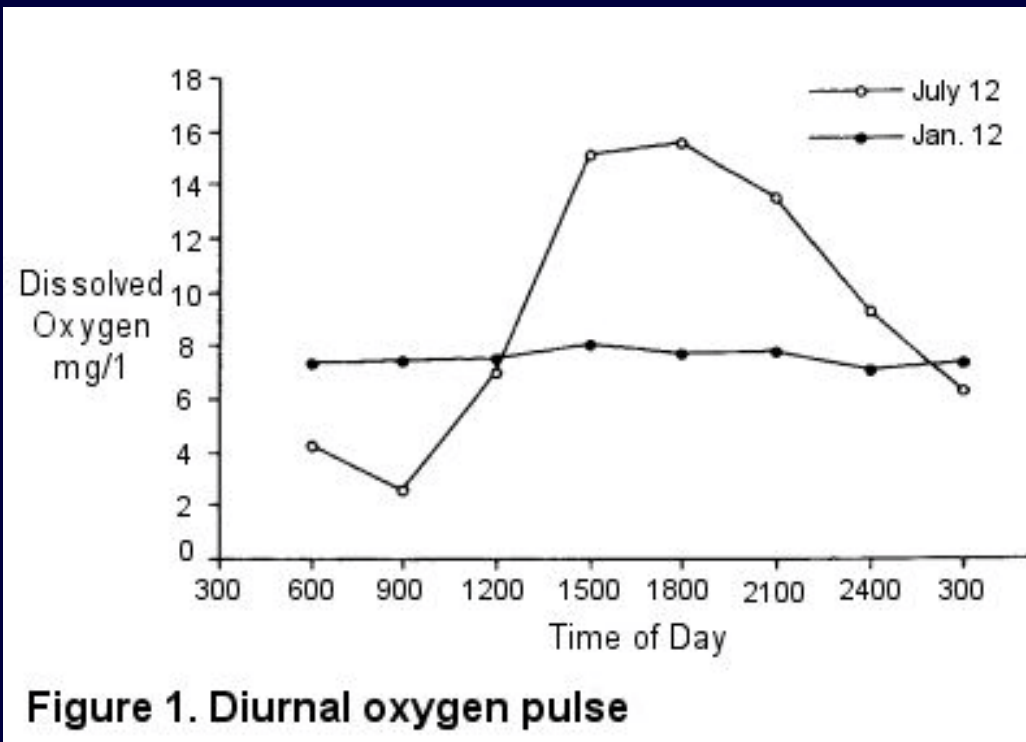
Fig. NAM-24-5 Seasonal isopleths of dissolved oxygen [ $\text{mg l}^{-1}$ ] in Gravenhurst Bay based on weekly measurements, May-October 1969, 1973 and 1974 (10).

# Fish Kills (low O<sub>2</sub>)



Also occur most often at dawn  
– why?

# At dawn



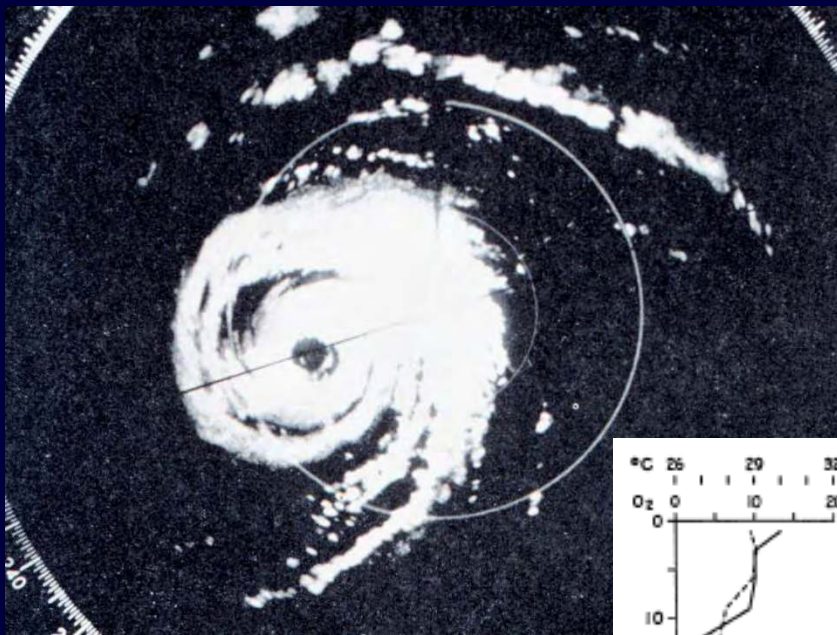
Eutrophic Timber Ridge Pond in OK, Freimuth and Bass 1994



# Fish Kills (low O<sub>2</sub>)



Also occur often after summer storm – why?



## Storm disruption of stratification

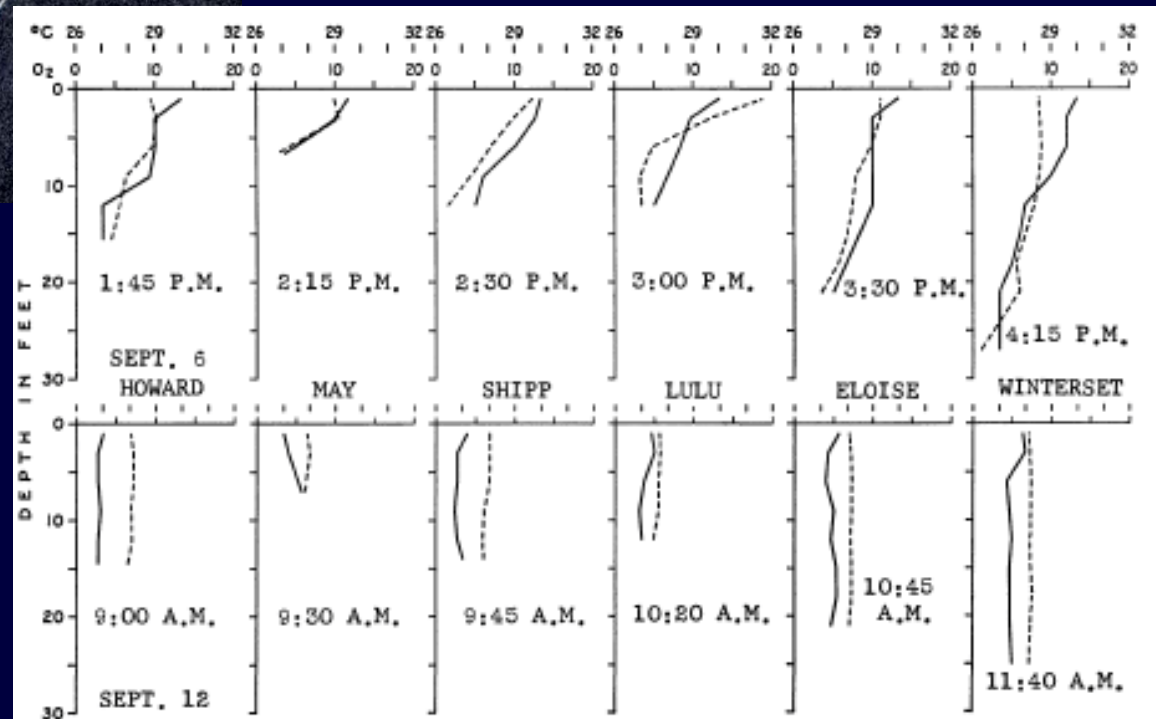


FIG. 2. Distribution of temperature (—) and oxygen (---) in Winter Haven lakes before and after hurricane "Donna."